

Képletek: Nernst-egyenlet:  $\Delta U = -\frac{RT}{zF} \cdot \ln\left(\frac{C_1}{C_2}\right)$   
 több ionra:  $\Delta U = -\frac{RT}{zF} \cdot \ln\left(\frac{P_{Na^+} \cdot [Na^+]_1 + P_{K^+} \cdot [K^+]_1 + P_{Cl^-} \cdot [Cl^-]_1}{P_{Na^+} \cdot [Na^+]_2 + P_{K^+} \cdot [K^+]_2 + P_{Cl^-} \cdot [Cl^-]_2}\right)$

P = permeabilitás

[x] = x koncentrációja

F = Faraday konstans = 96500  $\frac{C}{mol}$

Fick-egyenlet:  $j_d = -D \cdot \frac{dc}{dx}$

$D = \frac{R \cdot T}{f}$

$[j_d] = 1 \frac{mol}{cm^2 \cdot s}$

$[C] = 1 \frac{mol}{cm^3}$

$\left[\frac{dc}{dx}\right] = 1 \frac{mol}{cm^4}$

$[D] = 1 \frac{cm^2}{s}$

oszmózis nyomás:  $\Pi = R \cdot T \cdot \Delta C$

f = diffúzió akadályozó hatással rendelkező szűrő, membrán

$Q = CU$

$E = \frac{\Delta U}{d}$

$\Delta Q = C \cdot \Delta U$

$C = \epsilon_0 \epsilon_r \cdot \frac{A}{d}$

$\left[\frac{C}{A}\right] = \frac{1 \mu F}{cm^2}$  fajlagos kapacitás

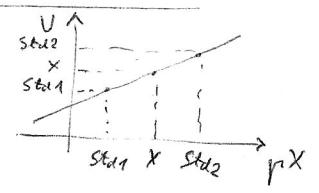
$\Delta U = U(I_1) - U(I_2)$

$\Delta Q = \left(\frac{C}{A}\right) \cdot A \cdot \Delta U$

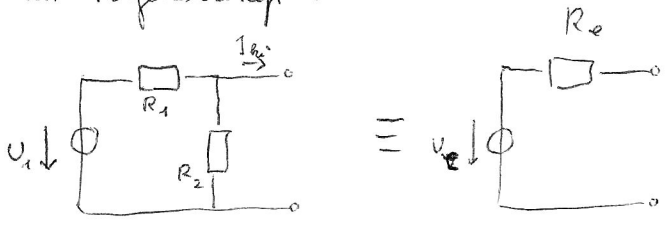
$Z = \frac{1}{j\omega C}$  ;  $X_C = \frac{1}{\omega C}$

vércs analízis:

$\frac{pX(x) - pX(Std1)}{U_{hi}(x) - U_{hi}(Std1)} = \frac{pX(Std2) - pX(Std1)}{U_{hi}(Std2) - U_{hi}(Std1)}$

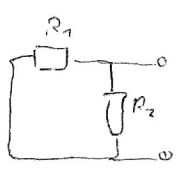


Thevenin-helyettesítéskép:



$U_{hi} \Big|_{I_{hi}=0} = U_1 \cdot \frac{R_2}{R_1 + R_2}$

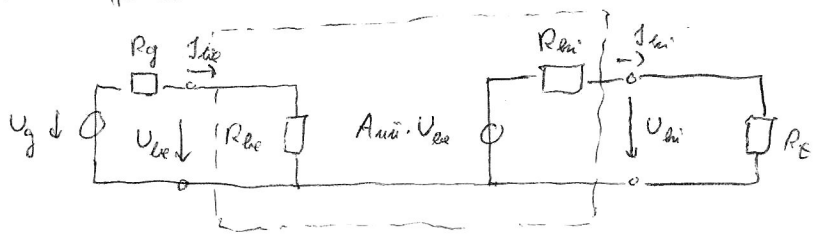
$\hookrightarrow U_e = U_1 \cdot \frac{R_2}{R_1 + R_2}$



$R_{le} = \frac{R_1 R_2}{R_1 + R_2} = R_e$

$\frac{1}{R_1} = \frac{1}{R_1} \parallel \frac{1}{R_2} \leftarrow \frac{U_e}{R_{le}}$

Erősítő felosztás:



$U_{be} = U_g \cdot \frac{R_{be}}{R_g + R_{be}}$

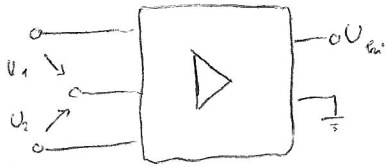
$R_{be} = \frac{U_{be}}{I_{be}}$

$U_{hi} = A_{nu} \cdot U_{be} \cdot \frac{R_t}{R_{in} + R_t}$

$R_{in} = -\frac{U_{hi} \text{ rögzítve}}{I_{hi} \text{ rögzítve}} \rightarrow [R_E \rightarrow \infty]$

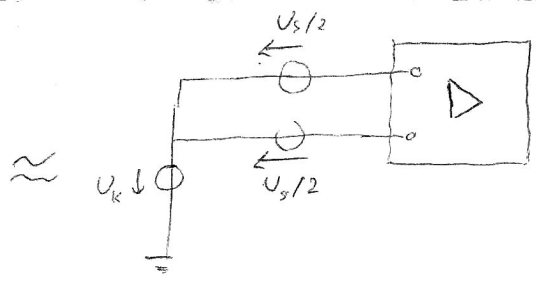
$\rightarrow [R_E = 0]$

Szimmetrikus erősítő :



$$U_S = U_1 - U_2$$

$$U_k = \frac{U_1 + U_2}{2}$$



$$U_{ki} = A_{u_s} \cdot U_{be_s} + A_{u_k} \cdot U_{be_k} = A_{u_s} \left( U_{be_s} + \frac{A_{u_k}}{A_{u_s}} \cdot U_{be_k} \right)$$

$$E_{ku} = \frac{A_{u_k}}{A_{u_s}} \quad \text{közösjelelőnyomás}$$

$$h_x = \frac{\partial F}{\partial x} \cdot \frac{\Delta x}{F}$$

Mechanikai értékelés :  $R = \rho \cdot \frac{l}{A}$

$$\Delta R = \frac{\partial R}{\partial \rho} \cdot \Delta \rho + \frac{\partial R}{\partial l} \cdot \Delta l + \frac{\partial R}{\partial A} \cdot \Delta A$$

$$\frac{\Delta R}{R} \approx \frac{\Delta \rho}{\rho} + \frac{\Delta l}{l} - \frac{\Delta A}{A}$$

$$(V = l \cdot A)$$

$$\frac{\Delta V}{V} = \frac{\Delta l}{l} + \frac{\Delta A}{A}$$

$$\frac{\Delta R}{R} \approx \frac{\Delta \rho}{\rho} + 2 \cdot \frac{\Delta l}{l}$$

$$G = \frac{\left(\frac{\Delta R}{R}\right)}{\left(\frac{\Delta l}{l}\right)} = 2 \quad \text{gauge-faktor}$$

1.  $P_K : P_{Na} : P_{Cl} = 1 : 0,03 : 0,1$

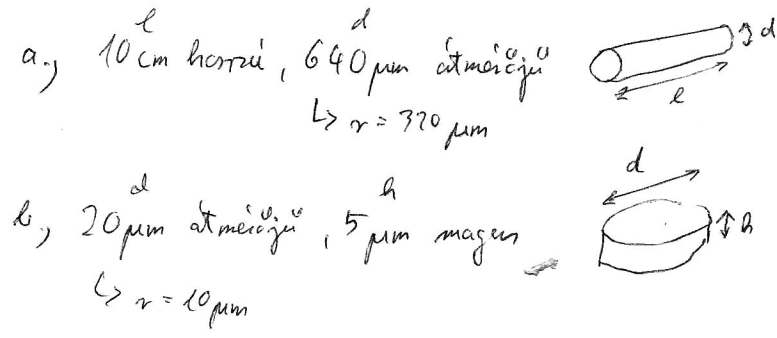
hínt	10	460	540
bent	400	50	40
ion	K	Na	Cl

megoldás:

$$\Delta U = -58 \text{ mV} \cdot \lg \left( \frac{1 \cdot 400 + 0,03 \cdot 50 + 0,1 \cdot 540}{1 \cdot 10 + 0,03 \cdot 460 + 0,1 \cdot 40} \right)$$

$$\Delta U = -70,44 \text{ mV}$$

2.  $\Delta U = +90 \text{ mV}$   
 $\frac{C}{A} = 1 \text{ } \mu\text{F/cm}^2$   
 sejtmembrán  
 lipíd réteg kapacitása  
 $Q = ?$



$F = 96500 \frac{\text{C}}{\text{mol}}$

megoldás:  $Q = C \cdot U$   
 $\Delta Q = C \cdot \Delta U = \left(\frac{C}{A}\right) \cdot A \cdot \Delta U$

$\Delta Q = 1 \frac{\mu\text{F}}{\text{cm}^2} \cdot A \cdot 90 \text{ mV} = 90 \cdot \frac{10^{-6}}{10^{-4}} \cdot 10^{-3} \cdot A \text{ [C]} = 90 \cdot 10^{-5} \cdot A \text{ [C]} = 900 \text{ A } \left[\frac{\mu\text{C}}{\mu\text{F}}\right]$

a.)  $A = 0,1 \cdot 2 \cdot 320 \cdot 10^{-6} \pi + \left(2 \cdot (320 \cdot 10^{-6})^2 \pi\right) \approx 0 = 201,062 \cdot 10^{-6} \text{ m}^2$   
 $= 201,062 \text{ } \mu\text{m}^2$

b.)  $A = 20 \cdot 10^{-6} \cdot 5 \cdot 10^{-6} + 2 \cdot (10 \cdot 10^{-6})^2 \pi = (100 + 400\pi) \cdot 10^{-12}$   
 $A = 9,425 \cdot 10^{-10} \text{ m}^2$   
 $A = 728,32 \cdot 10^{-12} \text{ m}^2 = 728,32 \cdot 10^{-6} \text{ } \mu\text{m}^2 = 728,32 \text{ } \mu\text{m}^2$

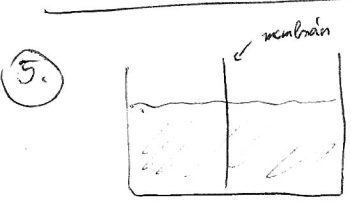
$\Delta Q_{a,y} = 900 \cdot 10^{-6} \cdot 201,062 \cdot 10^{-6} = 180,955,8 \cdot 10^{-12} \text{ [C]} = 180,9558 \cdot 10^{-9} \text{ [C]} \approx 180,96 \text{ pC}$   
 $\Delta Q_{b,y} = 900 \cdot 10^{-6} \cdot 728,32 \cdot 10^{-12} = 655,488 \cdot 10^{-18} \text{ [C]} = 655,488 \cdot 10^{-15} \text{ C} \approx 655,488 \text{ fC}$   
 $8,48 \cdot 10^{-13} \text{ C} \approx 0,85 \text{ fC}$  ✓

3. sejtmembrán vastagsága: 12 nm  
 $\epsilon_r = ?$   $\left(\frac{C}{A} = 1 \frac{\mu\text{F}}{\text{cm}^2}\right)$

$C = \epsilon_0 \epsilon_r \cdot \frac{A}{d} = 8,85 \cdot 10^{-12} \cdot \epsilon_r \cdot \frac{A}{12 \cdot 10^{-9}}$   
 $\frac{C}{A} = \frac{1 \cdot 10^{-6}}{10^{-4}} = 10^{-2} = 885 \cdot 10^{-12} \epsilon_r \cdot \frac{1}{12 \cdot 10^{-9}}$   
 $\epsilon_r = 13,56$

4.  $\Delta U = 70 \text{ mV}$   
 $d = 12 \text{ nm}$  }  $E = ?$

$E = \frac{\Delta U}{d} = \frac{70 \cdot 10^{-3}}{12 \cdot 10^{-9}} = 5,83 \cdot 10^6 \frac{\text{V}}{\text{m}} \approx 5833333 \frac{\text{V}}{\text{m}}$



a.) koncentrációkülönbség?  $\frac{C_1}{C_2}$   
 $\Delta U = -58 \text{ mV} \cdot \lg \frac{C_1}{C_2} \rightarrow \frac{C_1}{C_2} = 10^{\frac{+58 \text{ mV}}{-58 \text{ mV}}} = 0,6723 \cdot 1,4873 \approx 1,49$

b.)  $C_1 = 0,1 \frac{\text{mol}}{\text{l}}$  }  $C_2 = ?$   
 $h_T = \pm 1 \text{ k}$  }  $\Delta U = -\frac{RT}{zF} \cdot \ln \frac{C_1}{C_2}$   
 $(F = 96500 \frac{\text{C}}{\text{mol}}) \Rightarrow C_2 = 0,1 \frac{\text{mol}}{\text{l}} \cdot e^{-\frac{(-) 10 \text{ mV} \cdot 96500}{8,314 \cdot 298}} = \frac{C_1}{C_2}$   
 $C_{\text{max}} = 0,1 \frac{\text{mol}}{\text{l}} \cdot e^{-\frac{901 \cdot 8,314 \cdot 298}{96500}} = 0,099974 \frac{\text{mol}}{\text{l}}$   
 $C_{\text{min}} = 0,1 \frac{\text{mol}}{\text{l}} \cdot e^{-\frac{0,01 \cdot 8,314 \cdot 298}{96500}} = 0,099974 \frac{\text{mol}}{\text{l}}$   
 $1,72 \cdot 10^{-4} \% \text{ eltérés}$

KCl  
 $T = 298 \text{ K}$   
 $\Delta U = 10 \text{ mV}$   
 csak a K ionokra kell figyelni!

$$\Delta U = - \frac{RT}{zF} \ln \frac{C_1}{C_2} = - \frac{8,314 \cdot [298 \pm 1]}{(1) \cdot 96500} \cdot \ln \frac{C_1}{C_2} = 10 \text{ mV}$$

$$\Delta U = - \left( \frac{8,314 \cdot 298}{96500} \pm \frac{8,314}{96500} \right) \cdot \ln \frac{C_1}{C_2} = 10 \text{ mV}$$

$$\frac{C_1}{C_2} = \frac{10 \text{ mV}}{- \left( \frac{8,314 \cdot 298}{96500} \pm \frac{8,314}{96500} \right)} \Rightarrow C_2 = C_1 \cdot e^{\pm \frac{10 \text{ mV}}{\frac{8,314 \cdot 298}{96500} \pm \frac{8,314}{96500}}}$$

$$C_1 = 0,1 \frac{\text{mol}}{\text{l}}$$

$$C_2 = 0,1 \cdot e^{\frac{10^{-2}}{\frac{8,314 \cdot (298 \pm 1)}{96500}}} \Rightarrow \frac{C_{2 \text{ max}}}{C_{2 \text{ min}}} = \frac{e^{\frac{10^{-2}}{\frac{8,314 \cdot 297}{96500}}}}{e^{\frac{10^{-2}}{\frac{8,314 \cdot 299}{96500}}}} = \frac{e^{\frac{965}{8,314 \cdot 297}}}{e^{\frac{965}{8,314 \cdot 299}}}$$

$$\frac{C_{2 \text{ max}}}{C_{2 \text{ min}}} = e^{\left( \frac{965}{8,314 \cdot 297} - \frac{965}{8,314 \cdot 299} \right)} = 1,002617504 \approx 0,26175\% \text{ eltérés}$$

$$\approx \pm 0,13\% \Rightarrow 0,1 \frac{\text{mol}}{\text{l}} \cdot 0,13\% \cdot \frac{1}{1000} = 0,13 \frac{\text{mmol}}{\text{l}}$$

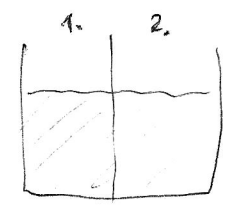
6.

Na-na ionváltási elektrod:

- $P_{Na} = 2 \cdot 10^{-4} \frac{\text{cm}}{\text{s}}$
- $P_K = 10^{-6} \frac{\text{cm}}{\text{s}}$
- $P_{Cl} = 5 \cdot 10^{-7} \frac{\text{cm}}{\text{s}}$
- $T = 25^\circ\text{C}$

	1.	2.
$K^+$	20 mM	400 mM
$Na^+$	20 mM	40 mM $\rightarrow$ 48 mM
$Cl^-$	100 mM	50 mM

(M = mol)



a.)  $\Delta U = ?$

$$\Delta U = -58 \text{ mV} \cdot \lg \frac{10^{-6} \cdot (20) + 2 \cdot 10^{-4} \cdot (20) + (50) \cdot 5 \cdot 10^{-7}}{10^{-6} \cdot 400 + 2 \cdot 10^{-4} \cdot 40 + 100 \cdot 5 \cdot 10^{-7}} = 18,56 \text{ mV}$$

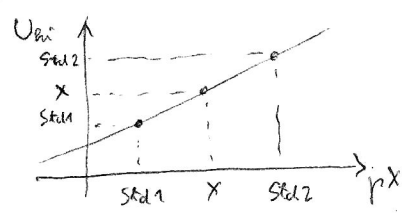
b.) a Na-koncentráció a 2-es kamrában megnövekedett 20%-kal. Hozzávaló elektrod az eredmény?

$$\Delta U = -58 \text{ mV} \cdot \lg \left( \frac{10^{-6} \cdot 20 + 2 \cdot 10^{-4} \cdot 70 + 50 \cdot 5 \cdot 10^{-7}}{10^{-6} \cdot 400 + 2 \cdot 10^{-4} \cdot 48 + 100 \cdot 5 \cdot 10^{-7}} \right) = 22,92 \text{ mV}$$

c.)  $[Na]_2 = 48 \text{ mM}$   
 $[K]_2 = 480 \text{ mM}$

$$\Delta U = -58 \text{ mV} \cdot \lg \frac{10^{-6} \cdot 480 + 2 \cdot 10^{-4} \cdot 48 + 500 \cdot 5 \cdot 10^{-7}}{10^{-6} \cdot 400 + 2 \cdot 10^{-4} \cdot 40 + 100 \cdot 5 \cdot 10^{-7}} = 23,12 \text{ mV}$$

7. Verigin - analysis



$pH(Std1.) = 6,838 \pm 0,005$   
 $pH(Std2.) = 7,382 \pm 0,005$   
 $U_{hi}(Std1.) = 1000 \text{ mV}$   
 $U_{hi}(Std2.) = 2000 \text{ mV}$

~~$pX(x) = \dots$~~   

$$\frac{pX(x) - pX(Std1)}{pX(Std2) - pX(Std1)} = \frac{U_{hi}(x) - U_{hi}(Std1)}{U_{hi}(Std2) - U_{hi}(Std1)}$$

~~a)  $U_{hi}(x) = 2061 \text{ mV}$~~

a)  $U_{hi}(x) = 2061 \text{ mV}$

$pX(x) = 6,838 + (2061 \text{ mV} - 1000 \text{ mV}) \cdot \frac{7,382 - 6,838}{2000 - 1000 [\text{mV}]}$   
 $pX(x) = 7,415$  ✓

b)  $U_{hi}(x) = 1900 \text{ mV}$

$pX(x) = 7,328$  ✓

c) min-max?

$$pX(x) = pX(Std1) + (U_{hi}(x) - U_{hi}(Std1)) \cdot \frac{pX(Std2) - pX(Std1)}{U_{hi}(Std2) - U_{hi}(Std1)}$$

$$pX(x) = pX(Std1) \cdot \left[ 1 - \frac{U_{hi}(x) - U_{hi}(Std1)}{U_{hi}(Std2) - U_{hi}(Std1)} \right] + \left[ \frac{U_{hi}(x) - U_{hi}(Std1)}{U_{hi}(Std2) - U_{hi}(Std1)} \right] \cdot pX(Std2)$$

b.) ~~min~~  $pX(x)$ , ha ~~min~~  $pX(Std1)$   
 or ~~min~~  $pX(Std2)$

a.) 1,061  
 b.) 0,9

~~max~~  $pX(x)$ , ha ~~max~~  $pX(Std1)$   
 or ~~max~~  $pX(Std2)$

a.) ~~min~~  $pX(x)$ , ha ~~max~~  $pX(Std1)$  or ~~min~~  $pX(Std2)$   
~~max~~  $pX(x)$ , ha ~~min~~  $pX(Std1)$  or ~~max~~  $pX(Std2)$  ✓

$$a.) pX(x)_{\min} = (6,838 + 0,005) \cdot (1 - 1,061) + (7,382 - 0,005) \cdot 1,061 = 7,4096$$
  

$$pX(x)_{\max} = (6,838 - 0,005) \cdot (1 - 1,061) + (7,382 + 0,005) \cdot 1,061 = 7,4208$$

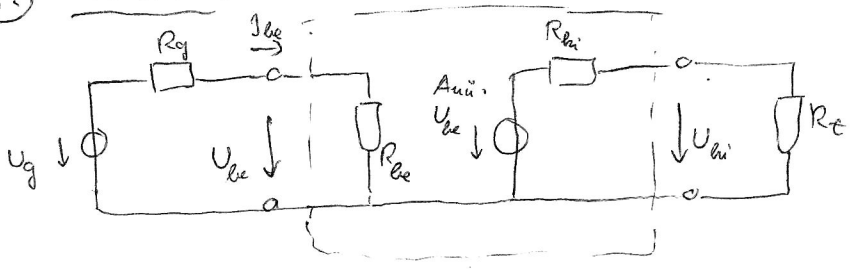
$$b.) pX(x)_{\min} = (6,838 - 0,005) \cdot (1 - 0,9) + (7,382 - 0,005) \cdot 0,9 = 7,3226$$
  

$$pX(x)_{\max} = (6,838 + 0,005) \cdot (1 - 0,9) + (7,382 + 0,005) \cdot 0,9 = 7,3326$$

a.)  $pX(x) = 7,410 \dots 7,421$  ✓

b.)  $pX(x) = 7,323 \dots 7,333$  ✓

8. exzista:



$$R_{be} = \frac{U_{be}}{I_{be}} \quad U_{be} = U_g \cdot \frac{R_{be}}{R_{be} + R_g}$$

$$R_{hi} = - \frac{U_{hi} \text{ "ausgangsspannung"}}{I_{hi} \text{ "einwärtelch"}} \quad U_{hi} = A_{ui} \cdot U_{be} \cdot \frac{R_L}{R_{hi} + R_L}$$

a.)  $R_L = \infty \rightarrow U_{hi} = 1V$   
 $R_L = 10k\Omega \rightarrow U_{hi} = 0,5V$   
 $R_{hi} = ?$

$$U_{hi_1} = A_{ui} \cdot U_{be} \cdot 1$$

$$U_{hi_2} = A_{ui} \cdot U_{be} \cdot \frac{10k\Omega}{R_{hi} + 10k\Omega} = \frac{U_{hi_1}}{2} \Rightarrow R_{hi} = 10k\Omega$$

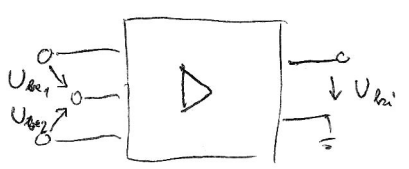
b.)  $R_g = 0 \rightarrow U_{hi_1} = 2V$   
 $R_g = 100k\Omega \rightarrow U_{hi_2} = 1V$   
 $R_{be} = ?$

$$U_{hi_1} = A_{ui} \cdot U_{be} \cdot \frac{R_L}{R_{hi} + R_L} = K_1 \cdot U_g \cdot \frac{R_{be}}{R_{be} + R_g} = 2V$$

$$U_{hi_2} = K_2 \cdot \frac{R_{be}}{R_{be} + R_g} = 1V$$

$$\frac{U_{hi_1}}{U_{hi_2}} = \frac{\frac{R_{be} + 0}{R_{be}}}{\frac{R_{be}}{R_{be} + 100k\Omega}} = 2 = \frac{R_{be}}{R_{be} + 100k\Omega} \Rightarrow R_{be} = 100k\Omega$$

9. symmetrischer exzista



$$U_S = U_{be1} - U_{be2} \quad \text{"j\u00f6"}$$

$$U_K = \frac{U_{be1} + U_{be2}}{2} \quad \text{"m\u00f6\u00dfer"}$$

Belastet:	$U_1 = 9,95V$	$U_1 = 5,1V$
$A_{uS}, A_{uK}?$	$U_2 = 10,05V$	$U_2 = 5V$
$E_{ku}?$	$U_{hi} = -0,9V$	$U_{hi} = 1,0505V$

$$U_{hi} = A_{uS} \cdot U_S + A_{uK} \cdot U_K$$

~~$$E_{ku} = \frac{U_K}{U_S}$$~~

$$E_{ku} = \frac{A_{uS}}{A_{uK}}$$

$$\Rightarrow -0,9V = -0,1A_{uS} + 10A_{uK} [V]$$

$$1,0505 = 0,1A_{uS} + 5,05A_{uK} [V]$$

$$\begin{cases} a.) -0,9 = A_{uS} \cdot (-0,1) + A_{uK} \cdot \frac{9,95 + 10,05}{2} \\ b.) 1,0505 = A_{uS} \cdot (+0,1) + A_{uK} \cdot 5,05 \end{cases}$$

$$A_{uK} = \frac{1,0505 + 0,1A_{uS}}{10}$$

$$-0,9 = A_{uS} \cdot (-0,1) + \left( \frac{1,0505}{10} + 0,01A_{uS} \right) \cdot 10$$

$$-0,9 = A_{uS} (1,0505 + 0,1 - 0,1) = 1,0505A_{uS}$$

$$A_{uS} = \frac{-0,9}{1,0505} = -0,8567$$

$$A_{uK} = 0,0965$$

$$\Rightarrow A_{uS} = 10; A_{uK} = 0,01; E_{ku} = 1000 \rightarrow E_{ku}^{dB} = 60dB$$